

# Ammonia (Natural Refrigerant)

Ammonia has been successfully used as a refrigerant in industrial refrigeration plants for over 130 years. It is a colourless gas, liquefies under pressure, and has a pungent odour. Ammonia has no ozone depletion potential (ODP = 0) and no direct global warming potential (GWP = 0). Thanks to its high energy efficiency, its contribution to the indirect global warming potential is also low. Ammonia is flammable and is toxic to skin and mucous membranes when highly pressurised. However, its ignition energy is 50 times higher than that of natural gas and ammonia will not burn without a supporting flame. Due to the high affinity of ammonia towards (air) humidity it is rated as "hardly flammable". Ammonia is toxic, but has a characteristic, sharp smell which makes a warning below concentrations of 3 mg/m<sup>3</sup> ammonia in air possible. This means that ammonia is evident at levels far below those which endanger health. Furthermore ammonia is lighter than air and therefore rises quickly.

## Characteristics:

|                                     |   |
|-------------------------------------|---|
| ODP                                 | 0   |
| GWP                                 | 0   |
| Appearance                          | colorless   |
| Odor                                | distinctive, biting   |
| Solubility in water (20 °C, 1 bar)  | 0.517 kg or 650 l(g)/l water  |
| Heat of solution                    | 36 kJ/mol   |
| Molecular weight                    | 17.03 kg/kmol   |
| Boiling point (1.013 bar)           | -33,3 °C  |
| Density of saturated vapour (20 °C) | 6.7025 kg/m <sup>3</sup>  |
| Thermic decomposition               | > 450 °C  |
| Explosibility limits                | 15 Vol.-% to 34 Vol.-%<br>108,000 mg/m <sup>3</sup><br>to 240,000 mg/m <sup>3</sup> |
| Ignition temperature                | 650 °C  |
| Ignition energy (20 °C, 101 kPa)    | 14 mJ   |
| Water content in the cycle          | of minor importance   |
| Perception threshold                | 5 ppm 3,5 mg/m <sup>3</sup>   |

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|--|---|
| MAK-Value  | 50 ppm 35 mg/m <sup>3</sup>               |
| Nuisance threshold                                 | 250 ppm 175 mg/m <sup>3</sup>             |
| Tolerance threshold                                | 500-1,000 ppm 350 -700 mg/m <sup>3</sup>  |
| Signs of poisoning                                 | 2,500 ppm 1,750 mg/m <sup>3</sup>         |
| Fatal dose   | > 5,000 ppm 3,500 mg/m <sup>3</sup>       |
| Long-term effects                                  | not carcinogenic, not mutagenic           |
| Concentration in human blood                       | 0,8-1,7 ppm                               |
| Daily production in the human body                 | 17 g ~ 1 mol                              |
| Water hazard class                                 | 2, ID No. 211                             |
| Evaporation enthalpy at 0 °C                       | 1,262 kJ/kg                               |
| Vapor pressure at 0 °C                             | 4.29 bar                                  |
| Compression ratio at 0 / 35 °C                     | 3.15                                      |
| Volumetric refrigerating capacity at 0 / 35 °C     | 3,798.2 kJ/ m <sup>3</sup>                |
| Isentropic refrigerating capacity number 0 / 35 °C | 6.75                                      |
| Isentropic final compression temperature 0 / 35 °C | 82.6 °C                                   |
| Heat conductivity of the liquid at 0 °C            | 518.5 * 10 <sup>-3</sup> W/mK             |
| Kinematic viscosity of the liquid at 0 °C          | 2.66 * 10 <sup>-7</sup> m <sup>2</sup> /s |
| Heat transition (evaporation, condensation)        | Very high                                 |